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Key indicators

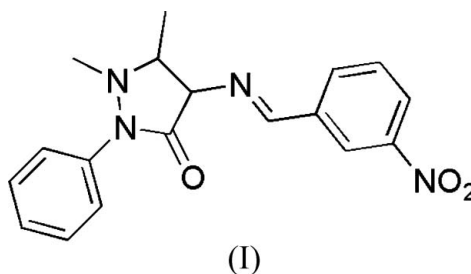
Single-crystal X-ray study
 $T = 294$ K
Mean $\sigma(\text{C}-\text{C}) = 0.005$ Å
 R factor = 0.062
 wR factor = 0.225
Data-to-parameter ratio = 12.6For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.1,5-Dimethyl-4-(3-nitrobenzylideneamino)-
2-phenylpyrazolidin-3-one

The title compound, $\text{C}_{18}\text{H}_{16}\text{N}_4\text{O}_3$, was prepared by the reaction of 3-nitrobenzaldehyde and 4-amino-1,5-dimethyl-2-phenylpyrazol-3-one. The crystal structure displays an intramolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen bond, which may stabilize the conformation of the molecule, as well as intermolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds.

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Comment

The syntheses and structures of Schiff bases have attracted much attention, because their metal complexes have been studied extensively as model compounds of biologically active compounds (Kahwa *et al.*, 1986; Santos *et al.*, 2001). As part of this investigation, we report here the synthesis and crystal structure of the title compound, (I).



A view of the molecular structure of (I) is shown in Fig. 1. The pyrazolidinone group C8–C10/N1–N3/O3 is planar, with an r.m.s. deviation for the fitted atoms of 0.0409 Å. This plane makes a dihedral angle of 48.55 (11)° with the phenyl ring (C13–C18) and a dihedral angle of 10.07 (11)° with the nitrobenzene ring (C1–C7/N4).

The intramolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen bond (Table 1), may help to stabilize the observed conformation of the molecule. In the crystal structure, the molecules are associated *via* weak $\text{C}-\text{H}\cdots\text{O}$ intermolecular hydrogen bonds (Table 1) to form a layer structure (Fig. 2).

Experimental

An anhydrous ethanol solution (50 ml) of 3-nitrobenzaldehyde (1.51 g, 10 mmol) was added to an anhydrous ethanol solution (50 ml) of 4-amino-1,5-dimethyl-2-phenylpyrazol-3-one (2.03 g, 10 mmol) and the mixture was stirred at 350 K for 5 h under nitrogen, whereupon a yellow precipitate was formed. The product was isolated and recrystallized from ethanol, and then dried *in vacuo* to give pure (I) in 83% yield. Bright-yellow single crystals of (I) suitable for X-ray analysis were obtained by slow evaporation of an ethanol solution.

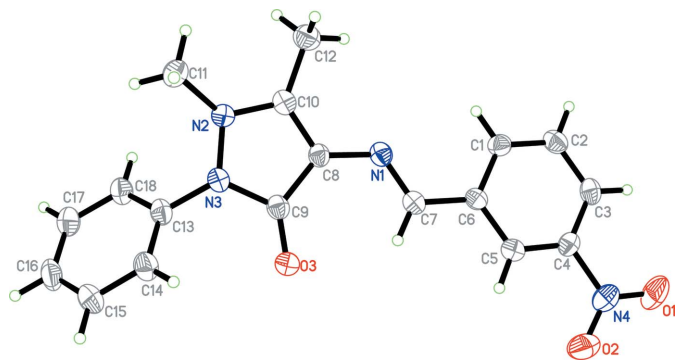


Figure 1
The structure of (I), with displacement ellipsoids drawn at the 30% probability level.

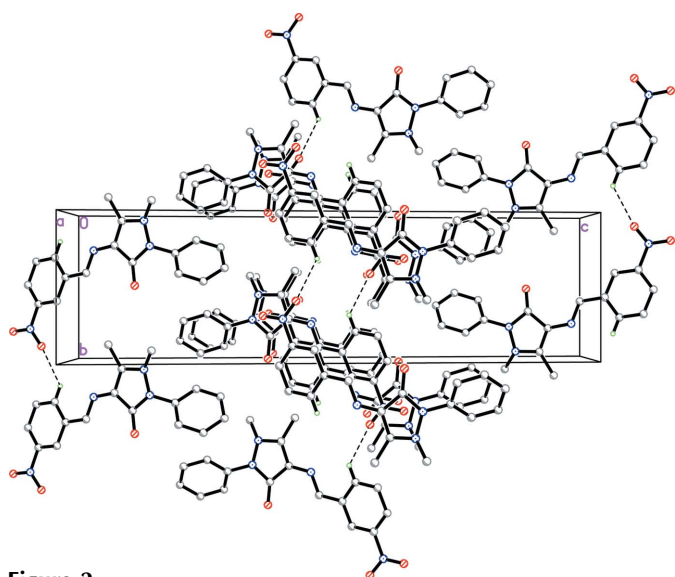


Figure 2
Intermolecular hydrogen-bonding interactions (dashed lines) in the crystal structure of (I). H atoms not involved in hydrogen bonds have been omitted.

Crystal data

$C_{18}H_{16}N_4O_3$
 $M_r = 336.35$
 Monoclinic, $P2_1/c$
 $a = 7.645$ (11) Å
 $b = 7.839$ (11) Å
 $c = 28.24$ (4) Å
 $\beta = 96.349$ (19)°
 $V = 1682$ (4) Å³
 $Z = 4$

$D_x = 1.328$ Mg m⁻³
 Mo $K\alpha$ radiation
 Cell parameters from 1975 reflections
 $\theta = 2.7$ – 25.8°
 $\mu = 0.09$ mm⁻¹
 $T = 294$ (2) K
 Block, yellow
 $0.32 \times 0.22 \times 0.16$ mm

Data collection

Bruker SMART CCD area-detector diffractometer
 φ and ω scans
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{min} = 0.962$, $T_{max} = 0.985$
 8161 measured reflections

2891 independent reflections
 1817 reflections with $I > 2\sigma(I)$
 $R_{int} = 0.044$
 $\theta_{max} = 25.0^\circ$
 $h = -6 \rightarrow 9$
 $k = -9 \rightarrow 9$
 $l = -33 \rightarrow 33$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.062$
 $wR(F^2) = 0.225$
 $S = 1.00$
 2891 reflections
 229 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.1519P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{max} = 0.001$
 $\Delta\rho_{max} = 0.30$ e Å⁻³
 $\Delta\rho_{min} = -0.27$ e Å⁻³
 Extinction correction: SHELXL97 (Sheldrick, 1997)
 Extinction coefficient: 0.008 (3)

Table 1

Hydrogen-bond geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
C7–H7 \cdots O3	0.93	2.45	3.094 (5)	126
C1–H1 \cdots O2 ⁱ	0.93	2.44	3.125 (6)	131

Symmetry code: (i) $x, y + 1, z$.

H atoms were included in calculated positions and refined using a riding-model approximation, with C–H = 0.93 Å and $U_{iso}(H) = 1.2U_{eq}(C)$ for aromatic CH, and C–H = 0.96 Å and $U_{iso}(H) = 1.5U_{eq}(C)$ for methyl CH₃.

Data collection: SMART (Bruker, 1999); cell refinement: SAINT (Bruker, 1999); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1999); software used to prepare material for publication: SHELXTL.

References

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